Name			Section		
				Date	
•					
EXP	PERIMENT 4	The D	etermina	tion of Boi	ling Point
PRE-	LABORATORY QUES	TIONS			
1.	Why is a <i>mixture</i> of ice and water, rather than ice alone, used in calibrating a thermometer?				
		4			
2.	Why does the boiling p	ooint of a liquid va	ary with the barome	etric pressure?	
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	4		•		
3.	Using a handbook of chemical data, find the boiling points (at 1 atm) for the following substances:				
	Acetone		,		
	Methyl ethyl ketone			-	
4	Methanol			7	
		,			
	Ethanol	T			

Reference